

## **Learning Journey – 7G Separating Mixtures**



What have I done previously in my learning journey?				
Previously	In year 5 you learnt:			
•	<ul> <li>How to compare and group together everyday materials on th including their hardness, solubility, transparency, conductivity and response to magnets.</li> <li>That some materials will dissolve in liquid to form a solution a a substance from a solution.</li> </ul>	(electri	cal and thermal),	
In this topic	We will learn about separation techniques and the affects on them, how to co	onduct a	n experiment to	
	compare solubility, what filtration is and how we use it, distillation, evaporation, chromatography as a separating technique, as well scientific skills such as planning a practical, considering variables in our planning and evaluating methods.			
We will develop our learning by studying the following each lesson:		RAG	Skills in Science checklist	
<ul> <li>Use particle m</li> </ul>	ixture is two or more atoms that are not chemically combined. odels to represent mixtures. eparation techniques are suitable, in terms of the properties of constituent		Scientific Method Practical Number skills Application Communication	
7G.02 Solutions			Scientific	
<ul> <li>State a solution contains dissolved particles</li> <li>Identify a solvent, solute, and solution in a given scenario</li> <li>Use the particle model to explain dissolving</li> <li>Draw particle diagrams to represent solutions and pure substances</li> </ul>			Method Practical Number skills Application Communication	
7G.03 Solubility Practical			Scientific	
<ul> <li>Describe how temperature affects solubility</li> <li>Explain why temperature affects the amount of solute dissolved in a solution</li> <li>Plan an investigation to compare solubility with temperature, considering variables</li> </ul>			Method Practical Number skills Application Communication	
7G.04 Solubility Practical			Scientific Method	
<ul> <li>Investigate how the temperature affects solubility</li> <li>Explain why some oceans in the world contain more salt than others</li> </ul>			Practical Number skills Application Communication	
7G.05 Filtration			Scientific	
<ul> <li>Draw a labelled diagram of the apparatus needed to filter a solution</li> <li>Explain how filtration works</li> <li>Explain whether or not filtering can be used in given situations</li> </ul>			Method Practical Number skills Application Communication	
7G.06 Evaporation and Distillation			☐ Scientific	
<ul> <li>Explain how di</li> </ul>	extures can be separated by evaporation stillation works ner evaporation or distillation would be suitable for separating different		Method Practical Number skills Application Communication	
7G.07 Chromatography			☐ Scientific	
<ul> <li>Analyse chrom</li> </ul>	to separate mixtures by chromatography latograms to identify substances in a mixture aromatography can be used in different scenarios		Method Practical Number skills Application Communication	
7G.08 Separating Seawater			☐ Scientific	
<ul> <li>Separate comp</li> </ul>	est way to separate seawater. conents of a mixture efficiently. method of separation.		Method Practical Number skills Application Communication	



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<ul> <li>7G.09 Separating Seawater Results</li> <li>Separate the components of a mixture using appropriate techniques</li> <li>Evaluate the efficiency of the techniques used by calculating the amount of salt, sand and water obtained.</li> </ul>	Scientific Method Practical Number skills Application Communication

Key Vocab	Solute: A substance that will dissolve in a liquid.		
	Solvent: The liquid that a solute substances dissolves in.		
	Solution: The product of a solute dissolving into a solvent.		
	Saturated: When a solute cannot dissolve anymore, and you can see it at the bottom of a beaker		
	Dissolve: When a solute (soluble substance) is added to a solvent (liquid that does the dissolving) to form a solution.		
	Particles: the smallest unit of matter that all substances are made from.		
	Soluble: If a substance can dissolve into a solvent, it is soluble.		
	Insoluble: If a substance cannot dissolve into a solvent, it is insoluble.		
Future Learning	At I/C4 consenting resistance is tought as next of shoresistance and usus vill leave research out should be resistant.		
Future Learning	At KS4 separating mixtures is taught as part of chemistry and you will learn more about chemical		
	analysis including separation techniques for mixtures of substances: filtration, crystallisation, chromatography, simple and fractional distillation		
In careers	Separation processes are essential to the chemical, petroleum refining, and materials processing		
	industries. The word "separation," however, refers to different processes and functions for		
	different industries. Separation processes comprise a large portion of the activity in the chemical and petrochemical industries.		